metal-organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Ming-Hua Zeng,^a Yang-Jian Hu^a and Seik Weng Ng^b*

^aDepartment of Chemistry, Guangxi Normal University, Guilin 541000, Guangxi, People's Republic of China, and ^bDepartment of Chemistry, University of Malaya, 50603 Kuala Lumpur, Malaysia

Correspondence e-mail: seikweng@um.edu.my

Key indicators

Single-crystal X-ray study T = 295 KMean $\sigma(C-C) = 0.006 \text{ Å}$ Disorder in solvent or counterion R factor = 0.052 wR factor = 0.161 Data-to-parameter ratio = 18.9

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

trans-Tetrachlorobis{1-[(2,6-diisopropylphenyl)iminiomethyl]-2-naphtholato- κO }zirconium(IV) dichloromethane disolvate

The Zr atom in the title disolvate, $[ZrCl_4(C_{23}H_{25}NO)_2]$ -2CH₂Cl₂, lies on a special position of $\overline{1}$ site symmetry in a *trans*-Cl₄O₂Zr octahedral geometry.

Received 7 August 2006 Accepted 9 August 2006

Comment

The deprotonated salicylideneamino-2,6-diisopropylbenzene Schiff base *N*,*O*-chelates to the metal atoms in a large number of metal complexes (Cambridge Structural Database, Version 5.27; Allen, 2002), an example of which is the dichlorozirconium(IV) complex (Strauch *et al.*, 2000). The isopropyl substituent of the diisopropylphenyl unit does not seem to be too sterically demanding in these examples. Similarly, the related Schiff base, 3-{[(2,6-diisopropylphenyl)imino]methyl}-2-naphthol (Lin *et al.*, 2005), is not expected to be crowded, and neither should the analogous Schiff base with an α -naphthyl group in place of the 2,6-isopropylphenyl unit, *i.e.* 3- $(\alpha$ -naphthyl)methyl-2-naphthol. Interestingly, in the solid state, 3- $(\alpha$ -naphthyl)methyl-2-naphthol exists in a zwitterionic form, as the hydroxy group has transferred its H atom to the imino group (Gavranic *et al.*, 1996).



 $\cdot 2CH_2Cl_2$

3-{[2,6-Diisopropylphenyl)imino]methyl}-2-naphthol, when reacted with zirconium(IV) tetrachloride, forms the title 1:2 metal-ligand adduct, (I), in which the ligand has also rearranged itself into a zwitterionic form so as to bind through the negatively-charged hydroxy O atom. The compound crystallizes from dichloromethane as a disolvate (Fig. 1). The metal atom lies on a special position of $\overline{1}$ site symmetry in a six-coordinate geometry. The bond dimensions of the iminium unit are similar to those found in, for example, zwitterionic 3-(α -naphthyl)methyl-2-naphthol (Gavranic *et al.*, 1996).

© 2006 International Union of Crystallography All rights reserved



Figure 1

A plot of zwitterionic (I). The disordered dichloromethane molecule is not shown. Displacement ellipsoids are drawn at the 50% probability level and H atoms are shown as spheres of arbitrary radii. The dashed lines indicate the intramolecular hydrogen bond. [Symmetry code: (i) = $\frac{1}{2} - x, \frac{1}{2} - y, 1 - z$.]

Experimental

The Schiff base was synthesized from 2-hydroxy-1-naphthaldehyde and 2,6-diisopropylaniline according to the literature procedure of Lin et al. (2005). The zirconium compound was synthesized using Schlenk techniques under a nitrogen atmosphere. The Schiff base (3.31 g, 10.5 mmol) was dissolved in tetrahydrofuran (THF; 100 ml) at 183 K. In an attempt to deprotonate the Schiff base, an n-butyllithium solution in hexane (0.683 g, 10.5 mmol) was added to the THF solution. The solution was allowed to warm to room temperature over a period of 2 h before being added to a solution of zirconium tetrachloride (1.17 g, 5.0 mmol) in THF (100 ml) at 183 K. The mixture was allowed to warm to room temperature. The solution was stirred overnight and then the solvent was removed under vacuum to give a solid, to which dichloromethane (60 ml) was added. The solid material that separated was collected and washed with dichloromethane. The solid material was discarded; the dichloromethane filtrate was concentrated to afford the crude product, (I), which was purified by recrystallization from dichloromethane to give yellow crystals in about 60% yield. CHN elemental analysis, calculated for C47H52N2O2Cl6Zr: C 57.55, H 5.34, N 2.86%; found: C 57.39, H 5.07, N 2.82%. These values are those of a mono- rather than a bis(dichloromethane) solvate.

Crystal data

$[ZrCl_4(C_{23}H_{25}NO)_2] \cdot 2CH_2Cl_2$	Z = 4
$M_r = 1065.75$	$D_x = 1.363 \text{ Mg m}^{-3}$
Monoclinic, $C2/c$	Mo $K\alpha$ radiation
a = 17.700 (1) Å	$\mu = 0.66 \text{ mm}^{-1}$
b = 16.897 (1) Å	T = 295 (2) K
c = 18.190 (1) Å	Block, yellow
$\beta = 107.286 \ (1)^{\circ}$	$0.25 \times 0.20 \times 0.16 \text{ mm}$
$V = 5194.6 (5) \text{ Å}^3$	

Data collection

Bruker APEXII area-detector	15120 measured reflections
diffractometer	5829 independent reflections
φ and ω scans	4410 reflections with $I > 2\sigma(I)$
Absorption correction: multi-scan	$R_{\rm int} = 0.018$
(SADABS; Sheldrick, 1996)	$\theta_{\rm max} = 27.5^{\circ}$
$T_{\min} = 0.852, \ T_{\max} = 0.902$	

Refinement

. .

7

(

 Refinement on F^2 $w = 1/[\sigma^2(F_o^2) + (0.0967P)^2$
 $R[F^2 > 2\sigma(F^2)] = 0.052$ + 2.9123P]

 $wR(F^2) = 0.161$ where $P = (F_o^2 + 2F_c^2)/3$

 S = 1.02 $(\Delta/\sigma)_{max} = 0.001$

 5829 reflections
 $\Delta\rho_{max} = 0.53$ e Å⁻³

 309 parameters
 $\Delta\rho_{min} = -0.36$ e Å⁻³

 H atoms treated by a mixture of independent and constrained refinement
 a^{-3}

Table 1	
Selected geometric parameters	(Å, °).

Cr1-O1 Cr1-Cl2	2.058 (2) 2.459 (1)	Zr1-Cl1	2.449 (1)
$\begin{array}{l} D1 - Zr1 - O1^{i} \\ D1 - Zr1 - Cl1 \\ D1 - Zr1 - Cl1^{i} \\ D1 - Zr1 - Cl2 \\ D1 - Zr1 - Cl2^{i} \end{array}$	180 89.09 (7) 90.91 (7) 87.02 (6) 92.98 (6)	$\begin{array}{c} Cl1\!-\!Zr1\!-\!Cl1^{i}\\ Cl1\!-\!Zr1\!-\!Cl2\\ Cl1\!-\!Zr1\!-\!Cl2^{i}\\ Cl2\!-\!Zr1\!-\!Cl2^{i} \end{array}$	180 89.28 (5) 90.72 (5) 180

Symmetry code: (i) $-x + \frac{1}{2}, -y + \frac{1}{2}, -z + 1$.

The dichloromethane molecule is disordered over two sites, the occupancies being in a 0.72 (1):0.28 (1) ratio. The C–Cl distance was restrained to 1.79 (1) Å and the Cl···Cl distance to 2.92 (1) Å. Additionally, the displacement parameters of the disordered atoms were restrained to be nearly isotropic.

The iminium H atom was located and refined with an N–H distance restraint of 0.85 (1) Å; its displacement parameter was freely refined. The remaining H atoms were placed in calculated positions (C-H = 0.93-0.98 Å) and were included in the refinement in the riding-model approximation, with $U_{iso}(H) = 1.2U_{eq}(C)$.

Data collection: *SMART* (Bruker, 2004); cell refinement: *SAINT* (Bruker, 2004); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *X-SEED* (Barbour, 2001); software used to prepare material for publication: *SHELXL97*.

The authors thank Guangxi Normal University, the National Science Foundation of Guangxi Province (grant No. 0447019) and the University of Malaya for supporting this study.

References

- Allen, F. H. (2002). Acta Cryst. B58, 380-388.
- Barbour, L. J. (2001). J. Supramol. Chem. 1, 189-191.
- Bruker (2004). SMART (Version 6.36a) and SAINT (Version 6.36a). Bruker AXS Inc., Madison, Wisconsin, USA.
- Gavranic, M., Kaitner, B. & Mestrovic, E. (1996). J. Chem. Crystallogr. 26, 23–28.
- Lin, J., Cui, G.-H., Li, J.-R. & Xu, S.-S. (2005). Acta Cryst. E61, 0627-0628.
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Strauch, J., Warren, T. H., Erker, G., Fröhlich, R. & Saarenketo, P. (2000). *Inorg. Chim. Acta*, **300**, 810–821.